States of Matter and Thermal Energy Reading Guide

Directions: Before you read your class assignment, read each statement in Part 1. If you believe that a statement is true, place a check in the *Agree* column. If you believe that a statement is false, place a check in the *Disagree* column. Be ready to explain your choices.

Agree	Disagree	
		1. All particles have kinetic energy because they are constantly moving.
		2. The particles in cold chocolate milk move faster than those in hot chocolate.
		3. Potential energy results from the attractive forces between particles.
		4. When an ice cube melts in your hand, the heat flows from the ice cube to your hand.
		5. The thermal energy of an object changes when its temperature changes.

Part 2:

Directions: Now you will read the information related to each of the statements in Part1. If the information is supported place a check in the *Yes* column in Part 2. Then <u>write what the text says in your own words</u> in column A. If the information does not support your choices, place a check in the *No* column. Then <u>write what the text says in your own words</u> in column B

Text to read: p. 254-259

	Support in text		A	В
	Yes	No	Why is my choice correct?	Why is my choice incorrect?
1.				
2.				
3.				
4.				
5.				