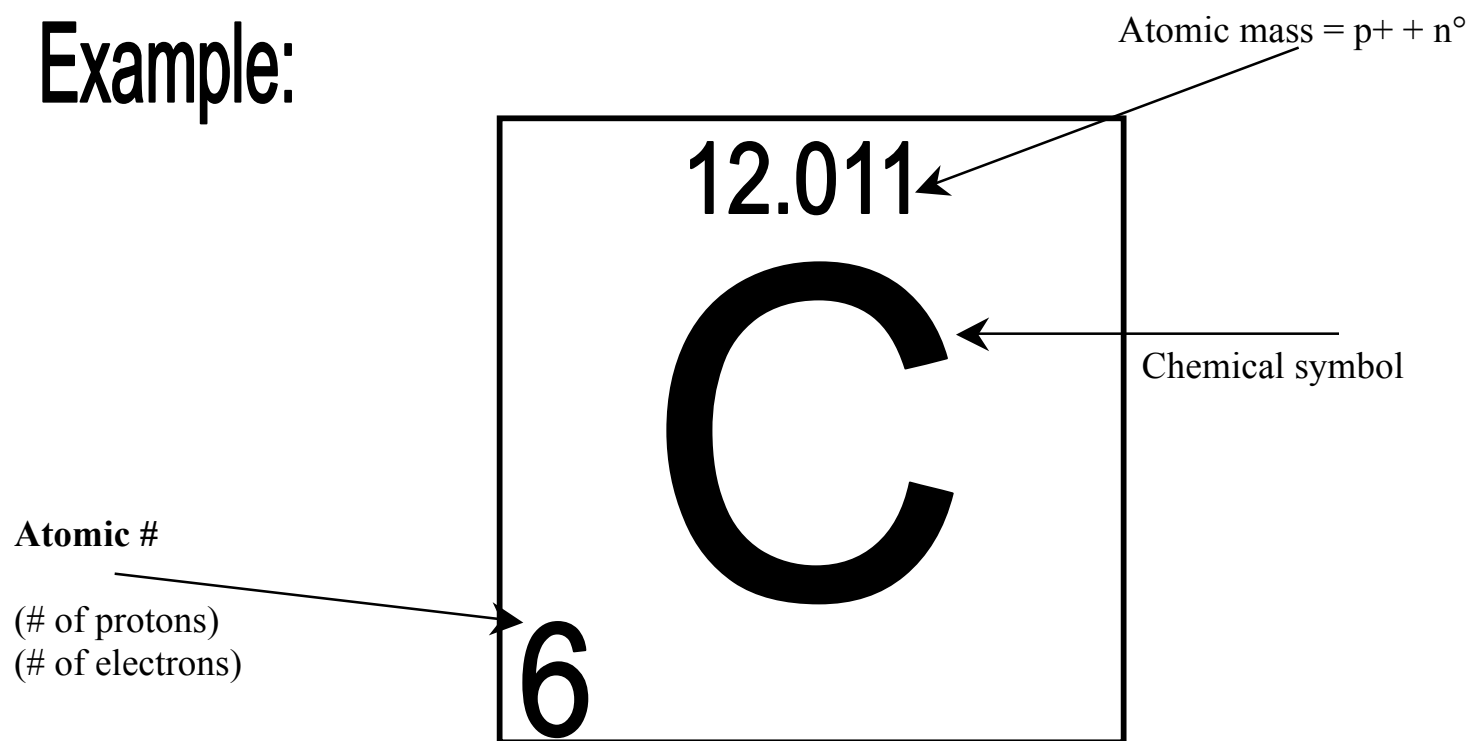


Symbol	Atomic Number	Mass Number	Number of Protons	Number of Electrons	Number of Neutrons
Na					
K		39		19	
			38	38	
F					
	20	40		20	
	50			50	
Si					
Kr					

Example:



$$\# \text{ Neutrons} = \text{Rounded Atomic Mass} - \# \text{ of protons}$$